Aqueous Diels-Alder Reactions of Cyclopentadiene with Symmetric Diester of Fumarate or Maleate

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Abstract : Symmetric diesters of *cis*- or *trans*- bicyclo[2,2,1]hept-5-ene-2,3-dicarboxylate were prepared by aqueous Diels-Alder reaction of cyclopentadiene with symmetric diester of fumarate or maleate.

Keywords: Aqueous Diels-Alder reaction, diester of bicyclo[2,2,1]hept-5-ene-2,3-dicarboxylate, diester of fumarate and maleate, cyclopentadiene.

Cis- or trans- dimethyl and diethyl ester of bicyclo[2,2,1]hept-5-ene-2,3-dicarboxylate (2 or 4) were used as pesticides $^{1\sim4}$ in the fourties and fifties of the century to prevent or kill mosquito, fly and flea. The cis-dimethyl ester 2a was also added to cosmetics for the same purpose 5 . The diethyl and dipropyl esters (2b and 2c) could be used to make rubber and elastomers $^{6.7}$. Besides, this kind of compounds are important intermediates for synthesis of some drugs.

The Diels-Alder reaction of cyclopentadiene with dienofiles 1 or 3 is the most simplest method for prepareing adducts 2 or 4. These methods are usually carried out in

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organic solvent , $\,$ most of them need catalyst . If the catalyst is absent, the reaction is sluggish 8 .

In 1980, Breslow reported an aqueous Diels-Alder reaction of cyclopentadiene with buten-2-one and revealed that the Diels-Alder reaction was accelerated by the hydrophoric effect of water ⁹. We have now tried the possibility of the aqueous Diels-Alder reaction of the symmetric diester of maleate (1) and fumarate (2) with cyclopentadiene. The reaction proceeded smoothly in an aqueous mixed solvent (water: acetone=3:2). The yields range from 74.7% to 94% after vacuum distillation.

Experimental

Typical procedure is examlified by the following reaction.

17.2 g (0.1 mol) of diethyl fumarate and 12.3 g (0.2 mol) of freshly distilled cyclopentadiene were added to 50 mL of solvent described above and the mixture was refluxed about 12 h. After separating the organic layer, the aqueous layer was extracted 3 times with petroleum ether, the combined organic phase were washed with water, dried over anhydrous magnesium sulphate, and evaporated. The residue was purified by vacuum distillation to give 21.18 g (89%) of product (4b), b.p. $106 \sim 108$ °C/1 mm¹⁰.

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- 10. **2a**, b. p. 94~96°C/2mm, yield 81. 5%; **2b**, b. p. 120°C/3mm, yield 86. 2%; **2c**, b. p 127~129 °C/3mm, yield 87. 5%; **4a**, b. p. 100~105°C, yield 94%; **4b**, b. p. 106 ∼ 108°C/1mm, yield 89%; **4c**, b. p. 122~128°C/1mm, yield 74. 7%.

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